## REACTION OF TROPONE WITH MESYLSULFENE J. Ciabattoni and M. Cabell<sup>\*</sup> Metcalf Research Laboratories, Brown University Providence, Rhode Island 02912

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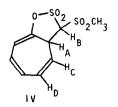
The relatively recent work by Opitz (1-4) on the generation and chemistry of mesylsulfene (11) and its stable trimethylamine adduct III prompts us to report our studies of the reaction

of tropone (5) with mesyl chloride and triethylamine. Dropwise addition (~1 hr) of a solution of purified triethylamine (33 mmoles) in anhydrous tetrahydrofuran to a stirred solution of tropone (10 mmoles) and mesyl chloride (30 mmoles) in anhydrous tetrahydrofuran at  $-5^{\circ}$ C under nitrogen resulted in the immediate precipitation of triethylamine hydrochloride. Filtration followed by solvent removal afforded after several recrystallizations from acetone a white crystalline compound<sup>\*\*</sup> in approximately 40% yield, m.p. 157-158°, <u>Anal</u>. Calcd. for C<sub>9</sub>H<sub>10</sub>S<sub>2</sub>O<sub>5</sub>: C, 41.21; H, 3.84; S, 24.45; mol. wt. 262. Found: C, 41.32; H, 3.97; S, 24.18; mass spectral data: <sup>m</sup>/e 262, 198, 184, 183, 182, 119, 118, 91, 90, 89, 79, 78, 77; osmometric mol. wt. in acetone 245;  $v_{max}^{KBr}$  (cm<sup>-1</sup>) 3050 (w), 2930 (m), 1660 (m), 1385 (s), 1375 (s), 1311 (s), 1220 (m), 1196 (s), 1169 (m), 1131 (s), 1121 (s);  $\lambda_{max}^{C_2}H_5^{OH}$  (mµ) 270 (2460); nmr (60 Mc, acetone-d<sub>6</sub>) in ppm from tms, 6.10-6.75 (3H, complex multiplet), 5.96 (doublet, H<sub>B</sub>, J<sub>AB</sub> = 5.0 cps), 5.50 (quartet, H<sub>C</sub>, J<sub>AC</sub> = 4.0 cps, J<sub>CD</sub> = 9.6 cps) 3.50-3.75 (complex multiplet, H<sub>A</sub>), 3.46 (sing<sup>1</sup> C. CH<sub>3</sub>-).

The above data are clearly consistent with the apparent 2:1 sulfene-tropone adduct IV.

<sup>\*</sup> National Aeronautics and Space Administration Predoctoral Fellow.

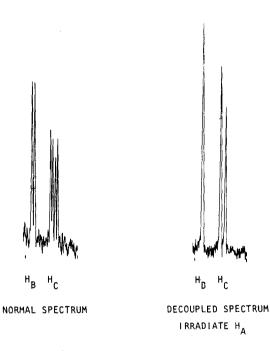
<sup>\*\*</sup> This compound can also be sublimed at 100<sup>0</sup>C (0.05 mm).



Addition of a trace of solid  $Na_2CO_3$  and a drop of  $D_2O$  to the nmr sample resulted in the complete disappearance of the doublet at 5.96 ppm as expected for the acidic hydrogen  $H_B$ .<sup>\*</sup> Double resonance data at 100 Mc served as additional proof of structure IV. Irradiation at 360 cps (3.60 ppm,  $H_A$ ) caused collapse of the doublet at 5.96 ppm ( $H_B$ ) to a singlet as well as the quartet at 5.50 ppm ( $H_C$ ) to a doublet,  $J_{CD} = 9.6$  cps (see Fig. 1).

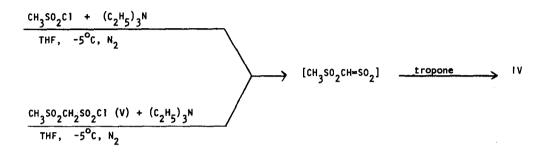
FIG. 1

DOUBLE RESONANCE DATA

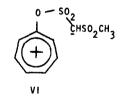


<sup>\*</sup> The Hitachi RMU-6D mass spectrometer as well as the Varian A-60A nmr spectrometer used in this work were purchased through a National Science Foundation major equipment grant to Brown University.

In an effort to understand the nature of the attacking species (1 or 11) in the above transformation, the reaction of tropone with mesylmesyl chloride (V) (6) with triethylamine was investigated. Reaction of tropone (5.2 mmoles) and mesylmesyl chloride (5.2 mmoles) in tetra-hydrofuran with triethylamine (5.7 mmoles) under nearly identical conditions as above afforded a white crystalline compound in approximately 32% yield, mp 157.5-159°, identical in all respects to the product of the reaction of tropone with mesyl chloride and triethylamine.



In view of the reported formation of II from either mesul chloride in acetonitrile at  $-40^{\circ}$ C or V in tetrahydrofuran at  $-70^{\circ}$ C (2-4), the intermediacy of II in our findings appears quite probable. The formation of IV can be rationalized by initial nucleophilic attack of the tropone oxygen on the sulfere sulfur of II to give the dipolar species VI followed by ring closure.



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## References

1. G. Opitz, Angew. Chem. Internat. Edit., 6, 107 (1967).

2. G. Opitz, M. Kleemann, D. Bucher, G. Walz, and K. Rieth, Angew. Chem., 78, 604 (1966).

3. G. Opitz and D. Bucher, Tetrahedron Letters, 5263 (1966).

4. G. Opitz, K. Rieth, and G. Walz, <u>Tetrahedron Letters</u>, 5269 (1966).

- For the reaction of tropone with various heterocumulenes see a) J. Ciabattoni and H. W. Anderson, <u>Tetrahedron Letters</u>, 3377 (1967).
  b) L. A. Paquette and N. Horton, <u>Tetrahedron Letters</u>, 0000 (1968).
  c) R. Gompper, A. Studeneer, and W. Elser, <u>Tetrahedron Letters</u>, 0000 (1968).
  d) I. Lengyel, R. Gompper, A. Studeneer, W. Elser, and Ch. Jutz, <u>Tetrahedron Letters</u>, 0000 (1968).
- 6. G. Opitz and G. Walz, unpublished results. We wish to thank Professor Opitz for a generous sample of compound V.